

Download Al₂S₃ Oxidation Number

The oxidation number of what in K₃PO₄? In that compound the oxidation number of oxygen is -2, the oxidation number of potassium is +1, and the oxidation number of phosphorus is +5. Atomic sulfur has oxidation number 0. The oxidation number of the sulfide ion is -2. During formation of a wide variety of compounds, the oxidation status of sulfur may differ from -2 to +6. Upload failed. Please upload a file larger than 100x100 pixels; We are experiencing some problems, please try again. You can only upload files of type PNG, JPG, or JPEG. Can someone please explain an easy way to find these oxidation numbers? MgO Al₂S₃ HNO₃ H₂SO₄ Ca(OH)₂ Fe(NO₃)₂